SESQUITERPENE AND DITERPENE LACTONES FROM MELAMPODIUM LONGIPILUM

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Abstract—The aerial parts of *Melampodium longipilum* afforded two new diterpene lactones, 17-hydroxy- and 17-acetoxyacanthoaustralide, and a new sesquiterpene lactone, repandin E, besides the known melampolides enhydrin and fluctuanin.

INTRODUCTION

The genera Melampodium, Enhydra, Polymnia, Tetragonotheca and Acanthospermum contain as characteristic constituents sesquiterpene lactones, mainly of the melampolide type [1]. Recent investigations of Acanthospermum australe [2] resulted in the isolation of novel diterpene lactones, which are most likely derived from geranylgeraniol by oxidative biomodifications. We describe in this paper the isolation and structure determination of the first two new members of this type of diterpene lactones from Melampodium species. Besides the two new diterpene lactones, 17-acetoxyacanthoaustralide (1b) and 17hydroxyacanthoaustralide (1a), a new sesquiterpene lactone (3) of the repandolide type [3] was also isolated from M. longipilum. In addition, the known melampolides enhydrin (4a) [4] and fluctuanin (4b) [5] were obtained from less polar fractions. Their identity was established by ¹H NMR spectral comparison with authentic samples. The co-occurrence of compound 3 with the two melampolide-4,5-epoxides is a further indication that the latter compounds represent biogenetic precursors of the repandolides, as we have previously suggested [3].

RESULTS AND DISCUSSION

17-Hydroxyacanthoaustralide (1a), $C_{20}H_{32}O_5$, was isolated as a gum from the more polar chromatography fractions. The IR spectrum indicated the presence of hydroxyl (3450 cm⁻¹) and carbonyl groups (1730 cm⁻¹). The ¹H NMR spectrum of 1a (Table 1) clearly indicated the presence of two vinyl methyl groups appearing as broad singlets at δ 1.72 and 1.77. A three-proton singlet at δ 1.34 suggested a tertiary methyl group on a carbon bearing an oxygen function. A doublet of doublets at δ 3.45 (J=11.0, 2.0 Hz), was assigned to a proton on a carbon bearing a secondary hydroxyl group. Two doublets of doublets centred at δ 4.18 and 4.00, which are the

Acetylation of 1a afforded a mixture of the tricetate 1c and the diacetate 1d which were separated by prep. TLC. The 1H NMR spectra of 1c and 1d showed minor but distinct differences. While the spectrum of 1c clearly indicated the presence of three sharp acetate peaks and a downfield shift of the H-1, H-6 and H-16 (or H-17) signals, the spectrum of 1d only exhibited two acetate signals and the downfield shift of the H-1 and H-16 (or H-17) signals. The presence of a secondary hydroxyl group at C-6 in 1d was indicated by a nearly unchanged doublet of doublets at δ 3.49 as well as the IR band at 3460 cm $^{-1}$.

From the less polar fractions, another diterpene lactone, 17-acetoxyacanthoaustralide (1b) was isolated. The IR and 1H NMR spectra of 1b were similar to those of 1a, indicating that the difference resided in the presence of an additional acetyl group in 1b (IR band at 1730 cm $^{-1}$) instead of a hydroxyl group in 1a. The acetate must be at C-17 since in the 1H NMR spectrum of 1b the C-17 two-proton methylene singlet was shifted downfield to $\delta 4.58$. This was in accord with the finding that acetylation of 1b gave the triacetate 1c which was identical with that obtained from 1a.

The stereochemistry at Δ^2 was assigned to have a Z-configuration based on the chemical shift of the C-3 methyl group (H-20) in **1a** and **1b**, which were in good agreement with the chemical shift of the C-3 methyl group of the Δ^2 (Z)-farnesols rather than the Δ^2 (E)-farnesols [6]. Oxidation of **1b** with manganese dioxide confirmed that the 2,3-double bond must have the Z-configuration, since the C-3 methyl group signal appeared at δ 1.99 in the

AB part of an ABX pattern, and an overlapped singlet at $\delta 4.07$ are due to the methylene protons of primary hydroxyl groups, since these signals shifted downfield upon acetylation. Three broad triplets at $\delta 5.23$ (J = 6.5 Hz), 5.54 (J = 7.0 Hz) and 6.35 (J = 5.0 Hz) indicated the presence of three vinyl protons, the latter possibly being β to a carbonyl group (IR band at $1700 \, \mathrm{cm}^{-1}$). All these ¹H NMR spectral features closely resembled those of acanthoaustralide (1g), a diterpene lactone recently isolated from A. australe [2]. The spectral data strongly suggested that 1g and the new compound (1a) differ by the presence of a hydroxyl function either at C-16 or C-17 in 1a. This was verified by the following chemical transformations.

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Table 1. ¹H NMR data for the diterpene lactones and derivatives (200 MHz, CDCl₃, TMS as int. standard, 27°)

Signals	1a	1 b	1c	1 d	2a	2b
H-1	4.0 dd (12, 7)*	4.01 dd (12, 7)	4.52 br d (7)	4.62 br d (7)	4.04 br d (7, 5)	4.05 (obs)
H-1'	4.18 dd (12, 8)	4.20 dd (12, 8)	_ ``	_ ` ` `		_
H-2	5.54 br t (7)	5.57 br t (7)	5.36 br t (7)	5.37 br t (7)	5.45 br t (7)	5.42 br t (7)
H-6	3.45 dd (11, 2)	3.46 dd (11, 2)	4.97 dd (11, 2)	3.49 dd (11, 2)	4.49 dd (9.5, 2.5)	4.5 dd (10, 2)
H-10	6.35 br t (5)	6.33 br t (5)	6.30 br t (5)	6.32 br t (5)	5.99 br t (4.5)	5.99 br t (4.5)
H-14	5.23 br t (6, 5)	5.37 br t (6, 5)	5.36 br t (9)	5.37 br t (7)	5.37 br t (6.5)	5.23 br t (6.5)
H-17	4.07 br s	4.58 br s	4.57 br s	4.57 br s	4.55 br s	4.05 br s
C-15-Me	1.77 br s	1.73 br s	1.73 br s	1.73 br s	1.73	1.73 br s
C-7-Me	1.34 s	1.34 s	1.35 s	1.35 s	1.26 s	1.24 s
C-3-Me	1.72 br s	1.73 br s	1.77 br s	1.76 br s	1.73	1.77
AcO	_	2.07 s	2.13, 2.07, 2.05 s	2.05, 2.06 s	2.07	****

^{*}Figures in parentheses are coupling constants or line separations in Hz.

3 R = Ac, R¹ = angelate, or vice versa 4a R = angelate epoxide

4b R = angelate

¹H NMR spectrum of the aldehyde 1e as in that of the Z-isomer in the spectrum of a citral mixture [7]. Although the ¹H NMR parameters for H-1, H-2 and the C-3-Me of compounds 1a and 1b were very similar to the chemical shifts of the reported acanthoaustralides [2], our assignments based on chemical shift data of aldehyde 1e are opposite to the configurational assignments reported in the literature [2].

The Δ^{14} -double bond configuration was assigned to be Z on the basis of the chemical shift of the aldehydic proton in the ¹H NMR spectrum of the aldehyde 1f obtained from 1a by oxidation with manganese dioxide, the chemical shift δ 10.08 being in agreement with a *cis*-aldehydic proton [8]. The stereochemistry at C-6 and C-7 remains open.

The ¹H NMR spectra of 1a and 1b indicated the presence of small amounts of another compound which must represent the isomeric eight-membered ring lactones 17-hydroxyisoacanthoaustralide (2a) and 17-acetoxy isoacanthoaustralide (2b). The assignments were based on the comparison of the ¹H NMR parameters with their analogues described in the literature [2]. Separation of compounds 1a and 2a by semi-preparative reverse phase HPLC on a 30 cm C-18 bondapak column (H₂O-MeOH, 1:1) gave baseline separations of the two lactones. However, after removal of the solvent mixture in vacuo, both fractions provided the same ratio of compounds 1a and 2a, as indicated by 200 MHz ¹H NMR. This strongly suggests that compounds 1a and 2a interconvert under the given conditions, most likely via intramolecular acyltransfer.

The structure of repandin E (3), $C_{23}H_{28}O_9$, mp 172–174°, was deduced mainly from the ¹H NMR and MS spectral data by comparison with the spectral parameters of the known repandins A–D [3]. The mass spectrum of repandin E showed the molecular ion at m/2 448. Further peaks at 388 [M – 60]⁺, 349 [M – 99]⁺, 83 [C₅H₇O]⁺ and 55 [C₄H₇]⁺ and diagnostic ¹H NMR signals indicated the presence of acetyl and angelyl ester side chains. The ¹H NMR spectrum also suggested that the two ester groups must be attached to C-8 and C-9, since the chemical shifts and couplings of the doublet of doublets at δ 6.21 (J = 10.3 Hz) and the doublet at 6.06 (J = 10 Hz), assigned to H-8 and H-9, respectively, were nearly identical with those of the known repandins [3]. These above ¹H NMR assignments and those of all other

protons were confirmed by spin-spin decoupling experiments of 3 (see Experimental). On the basis of the great similarity of the ¹H NMR and CD parameters of repandin E with the repandins A-D [3], it appears that the new compound has the configurations and conformation of the medium ring skeleton as previously established for the other repandolides [3]. The attachments of the two ester groups as shown in structure 3 with the acetate at C-9 and the angelate group at C-8 is solely based on the co-occurence of the known melampolide fluctuanin (4b), which appears to represent the biogenetic precursor for 3 [3]. Therefore, the sites of attachments of the two ester groups in 3 remain tentative.

EXPERIMENTAL

Melampodium longipilum Robins. was collected on 1 Sept. 1976 in Mexico: Puebla ca 30 miles north-west of Huajuapan de Leon on Highway 190. (Hartman & Funk No. 4152; voucher deposited at O.S., U.S.A.) Leaves and stems (49.0 g) were extracted with CHCl₃. Standard work-up [9] provided 2.0 g of crude syrup which was chromatographed on silica gel (80 g) using petrol-CHCl₃ (1:1), CHCl₃ and mixtures of CHCl₃-Me₂CO (5, 10, 20, 50%) as eluant; 100 ml fractions were taken and monitored by TLC. Earlier fractions eluted with petrol-CHCl₃ provided the sesquiterpene lactones enhydrin (4a) [4] and fluctuanin (4b) (longipin acetate) [5]. Their identity was established by ¹H NMR spectral comparison with authentic samples. Diterpene lactones 1a (200 mg) and 1b (160 mg) were obtained as gums from fractions eluted with CHCl₃-Me₂CO (19:1) and CHCl₃-Me₂CO (1:1).

17-Acetoxyacanthoaustralide (1b). $C_{22}H_{34}O_6$, gum; UV λ_{max}^{MeOH} nm(s): 202 (12 944), 220sh (7643); IR ν_{max}^{film} cm⁻¹: 3450, 1730, 1680, 1230; EIMS (probe) m/z (rel. int.): 350 [M – 44] + (0.6), 349 [M – 45] + (1.8), 348 [M – 46] + (1.6), 334 [M – AcOH] + (1.5), 316 [M – AcOH – H_2O] + (1.2), 265 [M – $C_7H_{13}O_2$] + (2.2), 205 [265 – AcOH] + (43.9), 167 (15.0), 151 (56.9), 149 (26.1), 147 (53.4), 133 (38.1), 121 (35.7), 119 (40.7), 93 (53.6), 91 (42.0), 83 (66.3), 81 (62.1), 43 (100.0).

17-Hydroxyacanthoaustralide (1a). $C_{20}H_{32}O_5$, gum; UV $\lambda_{\max}^{\text{MeOH}}$ nm(ϵ): 203 (11 776), 220 sh (4972); IR ν_{\max}^{film} cm $^{-1}$: 3450, 1690, 1620; EIMS (probe) m/z (rel. int.): 362 [M] $^+$ (not present), 279 (10.0), 205 (11.4), 167 (31.3), 151 (28.1), 149 (100.0), 133 (20.7), 121 (24.1), 119 (23.6), 109 (25.5), 107 (21.9), 105 (26.1), 95 (27.8), 93 (35.3), 91 (24.4), 83 (31.3), 81 (45.3), 79 (33.1), 71 (31.6), 69 (29.4), 67 (28.1), 57 (26.1), 55 (37.6), 53 (17.0), 43 (67.6), 41 (32.4).

Triacetate (1c) and diacetate (1d). Acetylation of 1b (50 mg) with Ac_2O (0.5 ml) and pyridine (0.2 ml) under standard work-up conditions gave after prep. TLC (Et₂O-petrol, 1:1, \times 4) 14 mg of 1c from the less polar band and 11 mg 1d.

Triacetate (1c). C₂₆H₃₈O₈, gum; UV λ $_{\rm max}^{\rm MeOH}$ nm(ε): 203 (9924), 220 (5873); IR $_{\rm max}^{\rm film}$ cm $^{-1}$: 1740 (esters), 1695 (conj. double bond), 1235 (acetate); EIMS (probe) m/z (rel. int.): 418 [M – AcOH] $^+$ (2.1), 358 [M – 2AcOH] $^+$ (1.6), 298 [M – 3AcOH] $^+$ (2.9), 279 (10.8), 205 (20.3), 164 (21.0), 167 (29.4), 151 (16.9), 149 (100.0), 133 (23.7), 121 (19.5), 119 (28.5), 109 (10.9), 107 (15.8), 105 (25.0), 95 (17.7), 93 (35.0), 91 (21.4), 83 (12.5), 81 (30.4), 79 (22.1), 71 (16.9), 69 (14.2), 67 (18.4), 57 (14.7), 55 (16.3), 43 (81.4).

Diacetate (1d). $C_{24}H_{36}O_7$, gum; UV λ_{max}^{McOH} nm(ϵ): 202 (8909), 220 sh (4739); IR ν_{max}^{film} cm⁻¹: 3460 (OH), 1740 (esters), 1700 sh, 1240 (acetate); EIMS (probe) m/z (rel. int.): 376 [M – AcOH]⁺ (1.4), 316 [M – 2AcOH]⁺ (2.8), 298 [M – 2AcOH – H_2O]⁺ (1.9), 205 (49.4), 164 (24.4), 151 (60.8), 149 (19.3), 147 (45.9), 133

(29.1), 121 (27.0), 119 (34.2), 109 (24.4), 107 (22.7), 105 (31.7), 95 (31.4), 93 (50.6), 91 (33.1), 83 (21.4), 81 (50.0), 79 (33.8), 71 (9.5), 69 (22.9), 67 (32.8), 55 (24.7), 43 (100.0).

Repandin E (3). C₂₃H₂₈O₉, was obtained from fractions eluted with CHCl₃-Me₂CO (19:1), mp 172-174° (CHCl₃-Et₂O); UV $\lambda_{\text{max}}^{\text{MeOH}}$ nm(ϵ): 216 (17309); CD (ϵ 4.7 × 10⁻⁵, MeOH): [θ]₂₁₄ - 141 072, [θ]₂₅₀ + 7 625; IR $\nu_{\text{max}}^{\text{fin}}$ cm⁻¹: 3550, 1770, 1740; EIMS (probe) m/z (rel. int.): 448 [M]⁺ (0.2), 388 [M - AcOH]⁺ (3.9), 349 (M - AngO]⁺ (4.6), 348 [M - AngOH]⁺ (1.4), 288 [M - AcOH - AngOH]⁺ (1.8), 83 [C₅H₇O]⁺ (100.0), 55 [C₄H₇]⁺ (26.1); ¹H NMR (CDCl₃): δ 6.75 (dd, J = 2 Hz, H-13b), 6.21 (dd, J = 10, 3 Hz, H-8), 6.06 (d, J = 10 Hz, H-9) 6.05 (qq, J = 7, 1.5 Hz, H-3'), 5.87 (d, J = 2 Hz, H-13a), 4.99, 5.01 (br s, H-15a, H-15b), 4.83 (br d, J = 9.5 Hz, H-6), 3.92 (d, J = 9.5, H-5), 3.94 (s, CO₂Me), 3.01 (m, H-7), 1.95 (s, OAc), 1.92 (dq, J = 7, 1.5 Hz, H-4'), 1.80 (quint, J = 1.5, H-5').

Aldehyde 1e. A 20 mg sample of 1b in 3 ml of CH₂Cl₂ was oxidized with MnO₂ the reaction being monitored by TLC. Usual work-up gave the gummy aldehyde 1e (14 mg). IR $v_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3500, 1735, 1680; EIMS (probe) m/z (rel. int.): 392 [M]⁺ (8.1), 332 [M - AcOH]⁺ (30.0), 314 [M - AcOH - H₂O]⁺ (4.6), 265 [M - C₇H₁₂O₂]⁺ (8.7), 205 [265 - AcOH]⁺ (47.1) 167 (100.0), 149 (29.9), 147 (28.1), 133 (30.8), 121 (28.8), 119 (19.7), 109 (78.8), 105 (25.3), 95 (30.9), 93 (35.0), 91 (21.7), 83 (26.9), 81 (93.0), 79 (27.2), 71 (18.7), 69 (62.0), 67 (29.7), 55 (23.3), 43 (69.7); ¹H NMR (CDCl₃): δ 9.95 (s, H-1), 6.34 (br t, J = 4.5 Hz, H-10) 5.92 (br d, J = 7.0 Hz, H-2), 5.37 (br t, J = 6.5 Hz, H-14) 4.57 (br s, H-17), 3.53 (dd, J = 11.0 Hz, J = 2.0 Hz, H-6) 2.07 (s, OAc), 1.99 (d, J = 1.0 Hz, H-20), 1.74 (br s, H-16), 1.35 (s, H-19).

Aldehyde 1f. MnO₂ oxidation of 18 mg of 1a gave after prep. TLC, the aldehyde 1f (8 mg) and a small amount of the C-1 monoaldehyde derivative (1h) which could not be separated due to the lack of material. IR $v_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 3400, 1680; ¹H NMR (CDCl₃): δ 10.08 (s, H-17), 6.48 (br t, J = 7.0 Hz, H-14), 6.36 (br t, J = 5.0 Hz, H-10), 5.58 (br t, J = 7.0 Hz, H-2), 4.22 (dd, J = 12.0 Hz; J = 8.0 Hz, H-1'), 4.02 (dd, J = 12.0 Hz; J = 7.0 Hz, H-1), 3.47 (dd, J = 11.0 Hz, J = 2.0 Hz, H-6), 1.78 (br s, H-16), 1.74 (br s, H-20), 1.34 (s, H-19).

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REFERENCES

- Seaman, F. C., Fischer, N. H. and Stuessy, T. F. (1980) Biochem. Syst. Ecol. 8, 263.
- Bohlmann, F., Jakupovic, J., Dhar, A. K., King, R. M. and Robinson, H. (1981) Phytochemistry 20, 1081.
- Seaman, F. C., Juneau, G. P., DiFeo, D. R., Jungk, S. and Fischer, N. H. (1979) J. Org. Chem. 44, 3400.
- Kartha, G., Go, K. T. and Joshi, B. S. (1972) J. Chem. Soc. Chem. Commun. 1327.
- Ali, E., Ghosh Dastidar, P. P., Pakrashi, S. C., Durham, L. J. and Duffield, A. M. (1972) Tetrahedron 28, 2285.
- Bates, R. B., Gale, D. M. and Gruner, B. J. (1963) J. Org. Chem. 28, 1086.
- 7. Sadtler catalog ¹H NMR spectrum No. 9477.
- Chan, K. C., Jewell, R. A., Nutting, W. H. and Rapaport, H. (1968) J. Org. Chem. 33, 3382.
- Fischer, N. H., Wiley, R. A., Lin, H. N., Karimian, K. and Politz, S. M. (1975) Phytochemistry 14, 2241.